

1. Matter

- 1.1. Introduction to matter
- 1.2. Classification of Matter
- 1.3. State of matter and phase changes
- 1.4. Properties of matter
- 1.5. Changes of matter
- 1.6. Elements
- 1.7. Compounds
- 1.8. Impure substances (Mixtures)
- 1.9 Assessment (True or false) with Solutions

2. Atomic Structure

- 2.1. Historical Development
 - i. J.J. Thompson
 - ii. Robert Millikan
 - iii. Ernest Rutherford
 - iv. Henry Mosley
 - v. Erwin Schrödinger
- 2.2. Quantum Numbers
 - i. Principal Quantum Number
 - ii. Azimuthal Quantum Number
 - iii. Magnetic Quantum Number
 - iv. Spin Quantum Number
- 2.2. Modern Ideas on Atomic Structure
- 2.3. Electronic Configuration
 - a) Sublevel Notation

i. Aufbau principle

ii. Pauli Exclusion Principle

iii. Hund's Rule of Maximum Multiplicity

b) Shell Notation

c) Electron Dot Notation (Lewis Dot Structure)

2.4. Assessment (True or false) with Solutions